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بسمه تعالی

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جناب آقای اسرارخانم دکتر زاهدانی

باسلام و احترام،

خوشایم اطلاع برسانم که مقاله اعمالات ارسالی شما علی بن دین کترش شیمی معدنی ایران با مشخصات زیر:

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Title: Synthesis and X-ray study of a new oxo-bridged heterotrinnuclear compound ...

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No:IICC-117

Title: Synthesis and Crystallography of a new oxo-bridge complex Cr, Mn with p-Chloro benzoate ligand

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No:IICC-120

Title: Synthesis and characterization analysis of new oxo-bridged, trimer of mixed-metal complexes with ...

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Title: Synthesis, characterization and IR investigation of novel oxo-centered, trinuclear of transition metal ...

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پس از دواوری توسط کمیته علمی برای ارائه کترش بصورت پوستر مورد پذیرش قرار گرفته اند.

لذا خواهشمند است جهت تکمیل مدارک زیر تا تاریخ ۱۳۸۷/۱۱/۲۰ اقدام فرمایید و اصل مدارک مربوط را با کترش خود و مقاله به دفتر دبیرخانه توسط پست ارسال فرمایید.

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Synthesis and characterization analysis of new oxo-bridged, trimer of mixed-metal complexes with acrylic acid

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Oxo-centered trinuclear mixed-metal complexes, $[M_2M'(\mu_3-O)(RCOO)_6(L)_3]^{n-}$ have played an important role in understanding the nature of metal-metal interaction, in particular magnetic properties of first-row transition elements [1-2]. The three ions are bound to central oxygen atom and adjacent metal ions are bridged by two carboxylate ligands. The monodentate ligand complete the octahedral coordination. These carboxylate complex have an additional interest when the carboxylate ligand is unsaturated, because there potential scope for further polymerization in the solid state by cross-linking of substituents [3].

In previous studies, mixed bridged trinuclear oxo-centered Iron(III), Chromium (III) : $[M_2M'(\mu_3-O)(\mu-O_2C_3H_3)_5(\mu-O_4C_6H_7)(O_2C_3H_3)(H_2O)_2]$ have been reported [4].

In this work symmetric species $[Fe_2Cr(\mu_3-O)(C_3H_3O_2)_6(H_2O)_3]^-NO_3^- \cdot 3H_2O(1)$ $[Cr_2CO(\mu_3-O)(C_3H_3O_2)_6(H_2O)_5]^+NO_3^- \cdot 3H_2O(2)$ have been prepared as NO_3^- salts.

These complexes have been characterized by elemental analyses, Infrared and electronic spectroscopy and DSC analyses. In the IR spectra of the complexes showed less than $200cm^{-1}$ differences between $\nu_{asym}(CO_2)$ and $\nu_{sym}(CO_2)$, indicating bidentate coordination of carboxylate group. DSC analysis shows that these crystals lose H_2O molecules in regions 86-115 °C and decompose completely in region 178-190 °C.

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