



دانشگاه سیستان و بلوچستان



بسمه تعالی

دهمین سمینار شیمی معدنی ایران- زاهدان- ۲۵ و ۲۶ اردیبهشت ماه ۱۳۸۷

10th Iranian Inorganic Conference (IICC-10), 14, 15 May 2008, Zahedan, IRAN

جناب آقای اسرارخانم دکتر زاهدانی

باسلام و احترام،

خوشایم اطلاع برسانم که مقاله اتمکات ارسالی شما برای دبیرخانه دین کتوش شیمی معدنی ایران با مشخصات زیر:

No:IICC-116

Title: Synthesis and X-ray study of a new oxo-bridged heterotrinnuclear compound ...

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No:IICC-117

Title: Synthesis and Crystallography of a new oxo-bridge complex Cr, Mn with p-Chloro benzoate ligand

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No:IICC-120

Title: Synthesis and characterization analysis of new oxo-bridged, trimer of mixed-metal complexes with ...

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No:IICC-121

Title: Synthesis, characterization and IR investigation of novel oxo-centered, trinuclear of transition metal ...

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پس از داوری توسط کمیته علمی برای ارائه کتوش بصورت پوستر مورد پذیرش قرار گرفته اند.

لذا خواهشمند است جهت تکمیل مدارک زیر تا تاریخ ۱۳۸۷/۱۱/۲۰ اقدام فرمایید و اصل مدارک مربوط را با کپی شده مقاله به دبیرخانه توسط پست ارسال فرمایید.

۱- تکمیل فرم ثبت نام (دریافت کتوش موجود است) (شماره ۱۱۷-۱۲۰)

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دانشگاه فردوسی مشهد

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d, Mashhad, Iran

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10th Iranian Inorganic Chemistry Conference
IICC-10, 2008



دانشگاه فردوسی مشهد

Synthesis and Crystallography of a new oxo-bridge complex Cr, Mn with p-Chloro benzoate ligand

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The Oxo-bridge complexes of transition metal with the general composition [M₂O(O₂CR)₆]^z (Fig.1) have been of interest for several years [1].



Fig.1. Structure of the cluster unit [M₂O(RCOO)₆(L)₃]^z

Where Met: Metal atom; RCOOH: carboxylic acid; L: terminal ligand such as water, methanol, pyridine (py), etc). z=+1 for M₃^{III} and z=0 for M₂^{III}M^{II} [2]. The preparation of oxo-bridge mixed metal complex [Cr₂MnO(C₇H₄O₂Cl)₆(Py)₃]·C₇H₅O₂Cl is described. The spectroscopic data in their solid states as well as in the dissolved form are reported for these complexes. This complex was characterized by X-ray crystallography, elemental analysis, atomic absorption, Middle and far IR and UV-VIS spectroscopy. The mixed-metal p-chloro benzoate complex presents a linear arrangement [3]. This complex crystallize isotypically in the monoclinic space group type P-1 with a= 26.2713(12) Å, b= 23.8658(11) Å, c =24.6895(11) Å, α =90°, β= 96.012(3) °, γ =90°, V=15394.8(12) Å³, Z= 8. The infrared spectra show resolved bands arising from ν_{asym}(OCO) and ν_{sym}(OCO) vibrations of monodentate and bridging carboxylate ligands along with those of ν_{asym}(M₂M'O) vibrations in the complex. The UV-VIS spectra exhibited two spin-allowed bands in the regions 440 and 558 (nm) which could be assigned to the transitions from ⁴A_{2g} to ⁴T_{2g} and ⁴A_{2g} to ⁴T_{1g} respectively. The spectra may be interpreted on the basis of an octahedral environment for chromium ion (III) (d³) in this complex [4].

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