

Effect of Pd as Dopant on Photocatalytic Microbial Inactivation of TiO₂ Thin Film



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Introduction

TiO₂ thin film has excellent photocatalyst properties. Because of that it has various applications such as air clean-up, water purification, anti-bacteria, antifogging coating, gas sensors and so on [1,2].

When TiO₂ is exposed to light, by the absorption of ultra band gap light it can be excited and generate electron/hole (e⁻/h⁺) pairs in the conduction/valence band region. The electron/hole pairs move to the surface and convert oxygen to superoxide radicals (O₂^{•-}) by reaction with the electrons and H₂O molecules to OH[•] radicals by reaction with the holes; then this radicals can participate in the photodegradation process and oxidize the pollutant [3,4].

In spite of beneficial applications, there are some limitations for using this semiconductor. One of them is the necessity of photo activation of this layer with UV illumination. Another is charge carrier recombination that results in decreasing photocatalytic activity. Some metals such as iron, cobalt, and chromium have been used as dopants to overcome these limitations. Presence of these metals in the film decreases the recombination of electron/hole and acts as charge separator. The excited electrons are transferred from the conduction band of the semiconductor to the metal, thus preventing recombination [5].

This film had been prepared by different kinds of deposition techniques such as sputtering, chemical vapor deposition, sol gel processing and so on. Sol gel techniques have an ability for the production of homogeneous layers at a molecular level, ease and versatility of processing and lower crystallization temperature [1,4].

In this current study, we used Palladium for improving antibacterial properties of TiO₂ thin film. We prepared the TiO₂ thin film loaded with Pd, using particulate sol gel route on stainless steel 316 substrates, then Evaluating antibacterial activity of coating against *S.aureus* bacteria was done by static antibacterial tests.

Experimental part

The solution for TiO₂ thin film was prepared by adding hydrochloric acid (HCl) 37% (Merck) to titanium tetraisopropoxide (TTIP) (Merck). Then distilled water was added dropwise to the solution under stirring. The resulting solution was stirred at 50°C for at least 1 hour. The final molar ratio of TTIP:HCl:H₂O was 0.1:0.2:53.3.

The preparation of Pd-TiO₂ sol was similar to that of TiO₂ sol. The only difference was that corresponding amount of palladium acetate (Merck) was added to TiO₂ sol to obtain 1%wt PdO in TiO₂ after calcination step.

Thin films of TiO₂ and Pd-TiO₂ were deposited over stainless steel 316. Plates were dipped into sols and after a few minutes withdrawn at the speed of 1mm/s. They were allowed to dry at room temperature and baked at 100°C for 10 min. Finally they were annealed at 550°C for 1 hour.

These thin films were characterized by means of X-ray diffraction (XRD) and scanning electron microscopy (SEM).

The antibacterial assessment was carried out to ensure the effect of palladium doping on semiconductor photocatalytic properties of TiO₂. *S.aureus* -- a gram-positive type of bacteria -- was cultivated in a nutrient broth (containing 10g/l beef extract, 10g/l peptone, 5g/l NaCl). Nano-structured samples were placed in a tube which contained solution. Then these tubes were put in the incubator, being exposed to UV light at 37°C.

Before coating the surface of stainless substrate with TiO₂, it was coated with a thin silica film. This thin film decreases the migration of substrate elements to TiO₂ coating that has a detrimental effect on photocatalytic activity of layer.

Results

FTIR analysis was used to identify the chemical changes which occurred in the solution. Wide and broad absorption peaks at about 576 Cm⁻¹ and 3323 Cm⁻¹ correspond to Ti-O-Ti and O-H surface vibration bands respectively. The absorption peak at about 1623 Cm⁻¹ is related to H-O-H stretching vibration bands (figure 1).



Figure 1: FTIR analysis of TiO₂ powder

XRD pattern of layer exhibited the presence of anatase crystalline phase. The peak at 2θ=25.4 corresponds to (101) reflections of the anatase phase, which has the greatest photocatalytic activity.

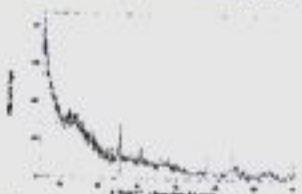


Figure 2: XRD pattern of TiO₂ coating at 550°C for 1 hour

Figure 3 illustrates SEM images of the layer that were annealed at 550°C for 1 hour. (a) represents the layer without palladium doping and (b) represents the layer with 1% palladium content. The presence of palladium is shown with white separated particles in TiO₂ film. The radius of Pd²⁺ is larger than Ti⁴⁺ (r_{Pd²⁺} = 0.8Å; r_{Ti⁴⁺} = 0.68Å), so it is unlikely that Pd²⁺ enter the TiO₂ lattice. Therefore Pd²⁺ is in direct contact with the TiO₂ nanoparticle.

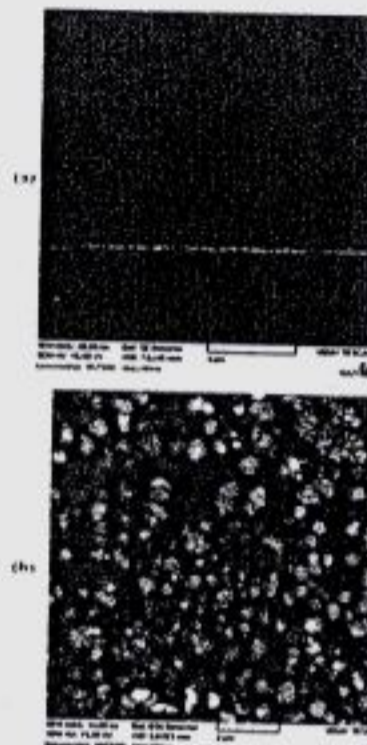


Figure 3. (a) SEM images of TiO₂ thin films (b) modified with 1% Pd content

Results of photocatalytic inactivation tests indicated that the samples which contained palladium show little photocatalytic activity without irradiation of UV light. These samples show high antibacterial activity for eliminating the *S.aureus* bacteria deposited after 2 hours' illumination with UV radiation. However, the relative number of bacteria which survived for TiO₂ coating is about 55% after 2 hours.

Conclusion

The doping with palladium enhances the photoactivation of semiconductor by UV. PdO-TiO₂ thin film indicated to be biocidally effective against *S.aureus* with a 100% kill in 2 hours.

Acknowledgements

The authors thank Mr Mohammadreza Rezaiyan for his help in preparing this poster.

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Certificate of Attendance

This is to certify that Mr/Mrs M. Rezaiyan participated in the 2nd International Congress on Nanoscience and Nanotechnology (ICNN2008), held by University of Tabriz in co-operation with the Iranian Nano Society (INS), on 28-30 October 2008, by presenting a paper entitled:

Effect of Pd as Dopant on Photocatalytic Microbial Inactivation of TiO₂ Thin Film

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