

Synthesis, characterization and crystal structure of Indium(III) binuclear complex containing pyridine-2,3 dicarboxylic acid

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The novel centrosymmetric complex, $[\text{In}_2(\text{pydc})_2(\text{pydcH})_2(\text{H}_2\text{O})_2] \cdot 2\text{H}_2\text{O}$ (1), (pydcH_2 = 2,3-pyridinedicarboxylic acid) was synthesized and thoroughly characterized using elemental analysis, IR, ^1H NMR spectroscopy and single crystal X-ray diffraction. All of obtained data via routine methods are in agreement with single crystal X-ray diffraction information. Compound 1 crystallized in the triclinic system with space group $\text{P}\bar{1}$ with two molecules per unit cell. Each In(III) is six-coordinated with InN_2O_4 bound set as distorted octahedral geometry. PydcH₂ with capability hydrogen donating and versatile coordination mode appears in this complex as both mono deprotonated, bidentate ligand and doubly protonated-tridentate one. Each tridentate ligands is coordinated to In(III) through two O atoms and one N atom and play as bridged role between two In(III) ions. The distance between two In(III) is about 6.527 Å and size of resultant hole within binuclear complex is around $3.195 \times 5.567 \text{\AA}^2$. In the title compound, different types of hydrogen bonds and electrostatic interaction play important roles in stabilizing the corresponding lattice and assemble to 3-D supramolecular network.

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