

Synthesis, Spectroscopic Characterization and Crystal Structure  
Determination of 3-(4-Chlorophenylthio)-1,3-diphenylpropan-1-one

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The conjugate addition of thiols to  $\alpha,\beta$  unsaturated carbonyl compounds leading to the formation of C–S bonds is a key reaction in the synthesis of organosulfur compounds as well as in biosynthesis. Organosulfur compounds have multiple applications in bioorganic chemistry and medicinal areas such as antibiotic, antimicrobial, anti-inflammatory, antitumor, and anti-HIV activities [1,2]. The title compound (Fig. 1) was prepared from the reaction of chalcone with 4-chlorobenzenethiol in the presence of lithium fluoride as nano-catalyst. This thiocarbonyl compound was characterized by elemental analysis, IR, UV-Vis,  $^1\text{H}$ NMR,  $^{13}\text{C}$ NMR and single-crystal X-ray determination. The title compound crystallizes in the space group  $P2_1/c$ , monoclinic, from  $\text{CH}_3\text{OH}/\text{CHCl}_3$ ,  $a = 19.638(3) \text{ \AA}$ ,  $b = 5.3734(11) \text{ \AA}$ ,  $c = 16.647(4) \text{ \AA}$ ,  $V = 1702.0(6) \text{ \AA}^3$ ,  $\beta = 104.336(19)^\circ$ ,  $\rho = 1.377 \text{ g/cm}^3$ ,  $Z = 4$ ,  $\mu = 0.35 \text{ mm}^{-1}$ ,  $F(000) = 736$ ,  $R_1 = 0.034$ ,  $wR_2 = 0.053$ ,  $\text{GOF} = 0.778$ , largest residual electron density peak/hole in the final difference map:  $0.24 / -0.23 \text{ e \AA}^{-3}$ . In general, all bond distances for the title compound are within the values characteristic according the atom involved, bond distances selected are: C(1)–C(2) 1.527(2), C(2)–C(3) 1.509(3), C(1)–C(10) 1.506(3), C(3)–C(4) 1.490(3), C(1)–S(1) 1.834(2), C(3)–O(1) 1.213(2) and C(16)–S(1) 1.772(2)  $\text{ \AA}$ . The asymmetric C atom has a slightly distorted tetrahedral configuration with the bond angles in the range of  $106.75(13)^\circ$  [S(1)–C(1)–C(2)] to  $113.32(17)^\circ$  [C(2)–C(1)–C(10)].

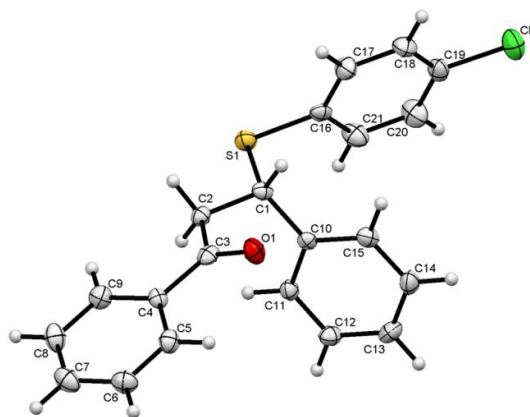


Fig. 1.

References:

- [1] M. K. Chaudhuri, S. Hussain, *J. Mol. Catal. A: Chem.* 269 (2007) 214.
- [2] Z. Xia, X. Lv, W. Wang, X. Wang, *Tetrahedron Lett.* 52 (2011) 4906.