Water-Driven Organic Syntheses Using Recyclable Nanocatalysts: Efficient

Routes to Diaryl Ethers, Aryl Ketones, and Triazoles

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This study highlights innovative approaches utilizing water as a green solvent in catalytic systems for sustainable organic transformations. Two heterogenous magnetic nanocatalysts, Fe₃O₄@Starch-Au and Fe₃O₄@HKUST-1 as well as one metal-organic framework (MOF) nanocatalyst, UIO-66/Sal-ZnCl₂, were developed. Characterization of the nanoparticles using FT-IR, XRD, SEM, TEM, EDS, VSM, and ICP techniques confirmed successful coating and stability. Diverse organic reactions were studied under the optimized conditions: Ullmann-type O-arylation of phenols and aryl tosylates to diaryl ethers,(1) hydration of terminal alkynes and nitriles to aryl ketones and amides,(2) azide-alkyne cycloaddition for 1,2,3-triazoles (3) and one-pot, three-component click reaction involving terminal alkynes, alkyl halides, and sodium azide to prepare 1,2,3-triazole heterocycles.(4) All systems emphasized water as a solvent, eliminating toxic reagents, organic solvents, and harsh conditions. The catalysts exhibited high efficiency, regioselectivity, magnetic recoverability, and underscoring their roles in eco-friendly protocols. These studies advance green chemistry by integrating aqueous media, recyclable nanocatalysts, and energy-efficient methodologies for scalable and environmentally benign synthesis.

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