Dynamic investigation of solid propellant combustion in a rocket motor chamber

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Abstract

This study considers dynamic investigation of combustion phenomena in a Solid Propellant Rocket Motor (SPRM). Gas dynamic in combustion chamber and effluent nozzle has been studies by solving governing equations (mass, energy and momentum balances). In the modeling, spread of flame and variation of mass added from propellant surface into the gas stream are considered. The governing equations, which were in Euler form, were solved numerically by applying Modified Mac-Carmak method. Pressure, temperature and velocity distributions were found as a function of time and axial distance. The results were validated by experimental data of a typical rocket motor. Effects of various factors on combustion were evaluated. The proposed model can be easily modified and used for other solid propellants even with low aluminum content.

Key words: rocket motor, combustion, propellant, dynamic modeling.

1. Introduction

Performance of a SPRM is rigorously depends on flow distribution, fuel composition and temperature distribution. These parameters can be specified by variables such as pressure, temperature, velocity, density, composition in the motor and variation of cross section area. In order to investigate effect of these variables on SPRM performance detailed model of the motor is necessary.

Several researchers have studied SPRM combustion modeling. Most of the earlier work in exploring the flow evolution in a rocket chamber was based on cold-flow studies with injection of inert gases through the sidewalls of the chamber simulating the gas influx from the burning propellant. In this idealized configuration, Taylor [1] and later Culick [2] obtained analytical expressions for the velocity distributions in laminar incompressible flows. The study was later extended by Balakrishnan et al. [3] to include the effects of rotationality and compressibility. Dunlap et al. [4] and Traineau et al. [5] conducted experiments for incompressible and compressible flows, respectively, to quantify the turbulence characteristics and mean velocity transitions in the chamber. Three regimes of flow evolution: the upstream laminar, transitional, and downstream turbulent regions, were identified. Beddini [6] and Sabnis et al. [7] applied turbulence closure models to obtain numerical results of the flow field.

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Sourabh et al [8] have simulated a SPRM by means of a large eddy simulation technique. They found that chemical reaction has laminate role in combustion at slower rate and propellant combustion maybe locally treated as a well stirred reactor. Pirotti et al [9] have concluded flame speeding fake in their one dimensional modeling. Their model is pure mathematical and haven’t been validated by experiments. Redulphe [10] have considered radiation in modeling of a SPRM. They have used Lagrangian approach to solve governing equations with CFD. Kim et al. [11] are divided a rocket motor into three region as solid phase, foam phase and gas phase. Conservation laws have been written for each section respectively. Jae-kun et al. [12] have employed one dimensional unsteady state models for combustion of RDX/GAP/BTTN in a rocket motor. They have adopted a rate of reaction as $r = k a^p$ from ref [13] in their modeling. They found that estimated burning rate of RDX/GAP/BTTN is in general higher that RDX/GAP alone. Their modeling responds in 1-100 atm pressure range.

The objective of current study is to provide a reliable and realistic model to analyze the combustion in SPRM. Considering erosive burning in modeling, generality of the proposed model and finally implementing artificial viscosity in numerical solution are the advantages of current article to previous researches.

2. Process Modeling

Figure 1, represents a typical SPRM chamber which is divided into three sections. Section between chamber front to $x=x_0$, entrance section, section between $x=x_0$ to $x=x_n$, propellant section, and finally nozzle section which is located between $x=x_n$ to $x=x_e$.

![Fig.1. Schematic of a typical SPRM.](image)

As hot gas enters the chamber, heats up the propellant. Propellant is ignited when a point on propellant surface reaches its critical ignition point [13]. This process continues when total amount of propellant is ignited. Because of high gas velocity and erosive burning there will be a great pressure in the chamber and by exiting gases from nozzle the ignition reaches a steady form.

2.1. Governing equations

In order to build a model for combustion phenomena the following assumptions have been used:

1. The flow is one-dimensional in the chamber. Turbulence of the gases in the chamber makes this assumption reasonable.
2. All the surface reactions take place in a thin zone near the surface (reaction surface).
3. Physical properties of igniter gas and the gases from propellant burning are the same.
4. Gases in the chamber obey perfect gas law.
5. Viscous dissipation and axial heat conduction are assumed to be negligible. the conservative laws are written as:

- **Continuity:**
  \[
  \frac{\partial (\rho A)}{\partial t} + \frac{\partial (\rho u A)}{\partial x} = r \rho_{pr} P_W 
  \]  \(1\)

- **Momentum:**
  \[
  \frac{\partial (\rho u A)}{\partial t} + \frac{\partial (\rho u^2 A)}{\partial x} + \frac{\partial (PA)}{\partial x} = P \frac{\partial A}{\partial x} - \tau_w P_W 
  \]  \(2\)

- **Energy :**
  \[
  \frac{\partial (EA)}{\partial t} + \frac{\partial [(E + P)u A]}{\partial x} = q_p - q_b 
  \]  \(3\)

Where

\[
q_p = r \rho_{pr} P_W h_f 
\]  \(4\)

\[
q_b = P_W h_c (T - T_{ps}) 
\]  \(5\)

Equation (1) – (3) are general form of conservative equations, but for each zone of chamber the right hand side of the equations have different forms as below:

**One)** \(x_i < x < x_n\)
- \(T_{ps} < T_{ps,ig}\)

In this case there is no ignition and the equations will have the following form:

- **Continuity:**
  \[
  \frac{\partial (\rho A)}{\partial t} + \frac{\partial (\rho u A)}{\partial x} = 0 
  \]  \(6\)

- **Momentum:**
  \[
  \frac{\partial (\rho u A)}{\partial t} + \frac{\partial (\rho u^2 A)}{\partial x} + \frac{\partial (PA)}{\partial x} = P \frac{\partial A}{\partial x} - \tau_w P_W 
  \]  \(7\)
\[ \frac{\partial (\rho A)}{\partial t} + \frac{\partial (\rho u A)}{\partial x} = \rho \rho_{pr} P_w \quad \text{(9)} \]

\[ \frac{\partial (\rho u A)}{\partial t} + \frac{\partial (\rho u^2 A)}{\partial x} + \frac{\partial (PA)}{\partial x} = P \frac{\partial A}{\partial x} - \tau_w P_w \quad \text{(10)} \]

\[ \frac{\partial (EA)}{\partial t} + \frac{\partial [(E + P)uA]}{\partial x} = q_p \quad \text{(11)} \]

Two) \( x_n < x < x_e \)

\[ \frac{\partial (\rho A)}{\partial t} + \frac{\partial (\rho u A)}{\partial x} = 0 \quad \text{(12)} \]

\[ \frac{\partial (\rho u A)}{\partial t} + \frac{\partial (\rho u^2 A)}{\partial x} + \frac{\partial (PA)}{\partial x} = P \frac{\partial A}{\partial x} - \tau_w P_w \quad \text{(13)} \]

\[ \frac{\partial (EA)}{\partial t} + \frac{\partial [(E + P)uA]}{\partial x} = 0 \quad \text{(14)} \]

### 2.2. Initial and boundary conditions

To solve equations (6)-(14) three initial conditions are necessary. These conditions which are the same for all location of chamber are:

\[ u(0,x) = u_i \quad \text{(24)} \]

\[ T(0,x) = T_i \quad \text{(25)} \]

\[ P(0,x) = P_i \quad \text{(26)} \]

Where \( u_i, T_i \) and \( P_i \) are initial velocity, temperature and pressure of the chamber. In order to avoid numerical difficulties \( P_i \) are assumed to be 0.0001 greater than atmospheric pressure. At exit section \( P_e \) is equal to ambient pressure and \( T(t,x_e) = T(t,x_e-dx) \), \( u(t,x_e) = u(t,x_e-dx) \). At interface sections Mach number, pressure and temperatures are equal. Boundary condition at \( x=x_0 \) are the same as exit section.
2.3. Auxiliary Equations

2.3.1. Heat transfer coefficient

Convective heat transfer coefficient has been calculated from Dittus-Bolter equation [14]. In this equation gas properties are calculated in average film temperature ($T_{af}$).

\[
h_c = 1.56 \times 10^{-3} \cdot \text{Pr}^{-0.6} \cdot C_p \left( \frac{\rho u}{R} \right)^{0.8} \cdot M_w^{0.1} \cdot T_{af}^{0.67} \cdot \left( \frac{x}{d} \right)^{-0.1} \cdot \left( \frac{A_1}{A} \right)^{0.4}
\]  

(27)

2.3.2. Friction factor

Colebrook correlation for turbulent flow has been used for friction factor calculation. Entrance region effects are also considered in this equation [15]. All properties in this equation are calculated at $T_{af}$.

\[
\frac{1}{f} = \frac{0.449 \cdot \left( \frac{d}{x} \right)^{0.1}}{\left\{ \ln \left[ \frac{(\varepsilon / d)}{3.7} + \frac{1.27 \cdot R \cdot M_w^{0.5} \cdot T_{af}^{1.65} \cdot \left( \frac{x}{d} \right)^{0.5}}{10^6 \cdot p \cdot u \cdot d \cdot f^{0.5}} \right] \right\}^2
\]

(28)

2.3.3. Burning rate

The burning rate has been adopted from Modified Lenoir-Robillard law [16]. In this equation effect of erosive burning is considered as:

\[
\dot{r} = a P^n + k h \exp \left( \frac{-\beta r \rho_{pr}}{\rho_{pr}} \right)
\]

(29)

$\beta$ was calculated experimentally from water-quench experimentation.

2.3.4. Propellant surface temperature before ignition

Propellant surface temperature before ignition increase, because of hot igniter gases. The temperature is described by the following equation by implementing heat conservation law across propellant thickness as:

\[
\frac{\partial T_{pr}(t, y)}{\partial t} = \alpha_{pr} \frac{\partial^2 T_{pr}(t, y)}{\partial y^2}
\]

(30)

Where

\[
T_{pr}(0, y) = T_i
\]

(31)

\[
T_{pr}(t, \infty) = T_{pi}
\]

(32)

\[
\frac{\partial T_{pr}(t, 0)}{\partial y} = \frac{hc(t)}{\lambda_{pr}} [T(t) - T_{pr}(t)]
\]

(33)

Equation (30) was solved by assuming a three order polynomial for $T_{pr}(y)$. The following ODE was obtained by inserting the polynomial.
\[
\frac{\partial T_{ps}}{\partial t} = \frac{4\alpha_{pr} h_{c}^2 (T - T_{ps})^3}{3\lambda_{pr} (T_{ps} - T_{pi})(2T - T_{ps} - T_{pi})}
\] (34)

In order to avoid singularity at time zero, initial condition for equation (34) is described as:

\[
T_{ps}(0) = T_{ps} + \varepsilon
\] (35)

Where \(\varepsilon\) is a small number like 0.1 K.

3. Numerical Solution

Equations (6)-(29) and equation (34)-(35) should be solved simultaneously. In order to solve the equations Modified Mac-Cormak (MMC) method has been used [17]. This method is a predictor corrector solution for the following equation:

\[
\frac{\partial U}{\partial t} + \frac{\partial E}{\partial x} = S
\] (36)

Where predictor equation is:

\[
U_{i+1}^{n+1} = U_{i}^{n} - \frac{\Delta t}{\Delta x} (E_{i+1}^{n} - E_{i}^{n}) + \Delta t \ast S_{i}^{n}
\] (37)

And corrector equation has the following form:

\[
U_{i}^{n+1} = \frac{1}{2} [U_{i}^{n} + U_{i+1}^{n+1} - \frac{\Delta t}{\Delta x} (E_{i+1}^{n+1} - E_{i}^{n+1}) + \Delta t \ast S_{i}^{n+1}] 
\] (38)

Equations (6)-(29) can be written in the form of equation (38) by defining \(U, S\) and \(E\) in the following vector forms:

\[
U = \begin{bmatrix} \rho A \\ \rho u A \\ EA \end{bmatrix}
\] (39)

\[
E = \begin{bmatrix} \rho u A \\ (u^2 \rho + P)A \\ (E + P)u A \end{bmatrix}
\] (40)

\[
S = \begin{bmatrix} \rho \frac{\partial A}{\partial x} - \tau_{w} P_{w} \\ q_{p} - q_{b} \end{bmatrix}
\] (41)

Vector \(S\) for each region of the chamber will have different values based on section 2.1. The following procedure is used to solve the equation according MMC method:

1. Using quantities values at time \(n\), the value of \(U, E\) and \(S\) at this time are known.
2. Use Eq. (38) to calculate \(U\) at time \(n+1\).
3. Using \(U_{i}^{n+1}\), calculate \(T, P, \rho\) and \(u\) at time \(n+1\).
4. Using quantities at time \(n+1\) and equation (40) and (41) calculate \(E\) and \(S\) at time \(n+1\).
5. Calculate \(U^{n+1}\) by replacing new quantities in Eq. (37).
6. Calculate quantities $T, P, \rho$ and $u$ at time $n+1$ from Eq. (39).

This procedure continues when the desired time is reached. In order to quarantine stability of the solution Mac Carmak and Baldwin have considered the following fourth order equation called Artificial Viscosity (AV):

$$\varepsilon \frac{\partial}{\partial x} \left[ u + a \frac{\partial^2 P}{\partial x^2} \right] + \frac{\partial u}{\partial x} = 0$$

(42)

The AV is added to numerical equation as:

$$U^n_i^n = U^n_i - \frac{\Delta t}{\Delta x} (F^n_i - F^n_{i-1}) + \Delta t \cdot S^n_i$$

(43)

$$U^{n+1}_i = \frac{1}{2} [U^n_i + U^{n+1}_i - \frac{\Delta t}{\Delta x} (F^{n+1}_i - F^{n+1}_{i-1}) + \Delta t \cdot S^{n+1}_i]$$

(44)

Where

$$F^n_i = E^n_i + A^n_i$$

(45)

$$F^{n+1}_i = \overline{E^{n+1}_i} + \overline{A^{n+1}_i}$$

(46)

$$A^n_i = \varepsilon [([u]_i^n + a^n_i) \cdot \left| \frac{P^n_{i+1} - 2P^n_i + P^n_{i-1}}{P^n_{i+1} + 2P^n_i + P^n_{i-1}} \right| (U^n_i - U^n_{i-1})]$$

(47)

$$A^{n+1}_i = \varepsilon [([u]_i^{n+1} + a^{n+1}_i) \cdot \left| \frac{P^{n+1}_{i+1} - 2P^{n+1}_i + P^{n+1}_{i-1}}{P^{n+1}_{i+1} + 2P^{n+1}_i + P^{n+1}_{i-1}} \right| (U^{n+1}_i - U^{n+1}_{i-1})]$$

(47)

For stability purposes $\varepsilon$ should be between 0 and 0.5. The AV is more important in the regions with pressure oscillation. Matlab version 7 has been employed to implement numerical routines.

4. Experimental Rig

In order to validate the model a typical jet motor was used. Table 1 shows the specification of the motor. Solid propellant physical properties are tabulated in Table 2.

| Quantity         | Value     |  |  |
|------------------|-----------|  |  |
| Entrance length  | 6.2 cm    |  |  |
| Fuel section length | 50 cm  |  |  |
| Cross section area of fuel section | 1.61 cm$^2$ |  |  |
| Wetted perimeter | 6.35 cm   |  |  |

Table 2. propellant physical properties (20% PBA-EPON, 80% AP(15μm 30%,180 μm 70%) )

<table>
<thead>
<tr>
<th>Quantity</th>
<th>Value</th>
</tr>
</thead>
<tbody>
<tr>
<td>$a=0.0005$</td>
<td>cm/s.cm</td>
</tr>
<tr>
<td>$n=0.4$</td>
<td></td>
</tr>
<tr>
<td>$\lambda_p=0.00377$</td>
<td>J/(cm.s.K)</td>
</tr>
<tr>
<td>$\rho=1.6$</td>
<td>gr/cm$^3$</td>
</tr>
<tr>
<td>$\alpha_p=0.001875$</td>
<td>cm$^2$/s</td>
</tr>
<tr>
<td>$M=2.01$</td>
<td>gr/gr-mol</td>
</tr>
</tbody>
</table>
5. Results and Discussion

Figures (3) and (4) show pressure variation with time for entrance and exit section of the chamber respectively. Plus sign (+) on pressure indicates that to avoid singularity in numerical solutions initial pressure is taken $P=1.0001$ atm. These figures show validity of predictions. Also they represent that erosive burning is important and should be considered in modeling. Perfect gas assumption seems main source of errors in the modeling and non ideality will be considered in the future works.

![Fig.3. Pressure variation with time for entrance section of the chamber.](image1)

![Fig.4. Pressure variation with time for exit section of the chamber.](image2)

Figure (5) represent pressure variation in combustion chamber during four different phases. At induction phase there is a good agreement between calculated and experimental results along the chamber. In flame propagation phase calculated pressure are a little bit less than experimental data. Maximum difference was observed for flame front. Calculated pressures are a little greater than experimental for chamber filling phase. In times near to steady state calculated pressures are the maximum attainable pressures in the chamber and they fit experiments.
Figure (5) shows pressure variation in the combustion chamber during four different regimes.

Figure (6) shows gas velocity variation with time for three different chamber zones. As gas velocity increases, ignition delay decreases. Figure (7) shows this fact. Increasing gas flow rate increases flame spreading time. Table 3 depicts this judgment.
Table 3. Effect of gas flow rate on delay time.

<table>
<thead>
<tr>
<th>Gas flow rate (gr/sec)</th>
<th>Ignition time (m sec)</th>
<th>End time (m sec)</th>
<th>Delay (m sec)</th>
</tr>
</thead>
<tbody>
<tr>
<td>15</td>
<td>38.49</td>
<td>59.1</td>
<td>20.16</td>
</tr>
<tr>
<td>20</td>
<td>23.51</td>
<td>38.32</td>
<td>14.81</td>
</tr>
</tbody>
</table>

Gas flow rate also affects pressure in the chamber. Figure (8) shows that by increasing gas flow rate:
- Induction pressure increases a little.
- In propagation phase this increase is sharp.
- Time increment from flame propagation to reach maximum pressure decreases.

Table 4 shows as the ratio of $A_p/A_t$ increases (by nozzle diameter) propagation phase starts earlier and flame spreading time decreases.

Table 4. Effect of nozzle diameter on flame spreading time and velocity.

<table>
<thead>
<tr>
<th>$A_p/A_t$</th>
<th>Ignition time (m sec)</th>
<th>End time (m sec)</th>
<th>Delay (m sec)</th>
<th>Initial flame spreading Velocity (cm/sec)</th>
<th>Final flame spreading Velocity (cm/sec)</th>
</tr>
</thead>
<tbody>
<tr>
<td>1.05</td>
<td>27.34</td>
<td>43.49</td>
<td>16.15</td>
<td>690</td>
<td>12500</td>
</tr>
<tr>
<td>1.20</td>
<td>23.56</td>
<td>38.28</td>
<td>14.72</td>
<td>800</td>
<td>14000</td>
</tr>
<tr>
<td>1.50</td>
<td>19.29</td>
<td>31.38</td>
<td>12.09</td>
<td>1020</td>
<td>16700</td>
</tr>
</tbody>
</table>

Fig.7. Dependency of ignition time with gas flow rate.

Fig.8. Dependency of Pressure with gas flow rate by the passage of time.

6. Conclusion and Remarks
In this article a model was developed to study solid-propellant jet motor by solving heat, mass and momentum conservation laws. In order to find the solution Modified Mac-Cormak method, which is stable for high pressure regions, with AV was applied to solve the equations. Results were validated with typical jet motor experimental data. Effect of operating parameters on combustion behavior was studied. The model is general and can be used for other solid propellants.

7. Nomenclature

<table>
<thead>
<tr>
<th>Symbol</th>
<th>Description</th>
</tr>
</thead>
<tbody>
<tr>
<td>$A_p$</td>
<td>Fuel section cross section area, cm$^2$</td>
</tr>
<tr>
<td>$A_t$</td>
<td>Nozzle cross section area, cm$^2$</td>
</tr>
<tr>
<td>$a$</td>
<td>Sound velocity, cm/sec</td>
</tr>
<tr>
<td>$C_p$</td>
<td>Heat capacity at constant pressure, erg/gr.K</td>
</tr>
<tr>
<td>$C_v$</td>
<td>Heat capacity at constant volume, erg/gr.K</td>
</tr>
<tr>
<td>$D$</td>
<td>Fuel section diameter, cm</td>
</tr>
<tr>
<td>$E$</td>
<td>Total energy, erg/gr</td>
</tr>
<tr>
<td>$e$</td>
<td>Internal energy, erg/gr</td>
</tr>
<tr>
<td>$f$</td>
<td>Friction factor</td>
</tr>
<tr>
<td>$h_c$</td>
<td>Convective heat transfer coefficient, erg/cm$^2$.sec.K</td>
</tr>
<tr>
<td>$k$</td>
<td>Coefficient of erosive burning rate, cm$^3$.K/cal</td>
</tr>
<tr>
<td>$m_{ig}$</td>
<td>Heating gas mass flow rate, gr/sec</td>
</tr>
<tr>
<td>$M$</td>
<td>Molecular weight, gr/gr-mol</td>
</tr>
<tr>
<td>$n$</td>
<td>Pressure power index in Eq. 29</td>
</tr>
<tr>
<td>$Pr$</td>
<td>Prandtel number, cm$^2$/sec</td>
</tr>
<tr>
<td>$P_w$</td>
<td>Perimeter of burning, cm</td>
</tr>
<tr>
<td>$q_b$</td>
<td>Heat transferred from gas stream to fuel, erg/cm$^2$.sec</td>
</tr>
<tr>
<td>$q_p$</td>
<td>Heat transferred from flame to gas stream, erg/cm$^2$.sec</td>
</tr>
<tr>
<td>$R$</td>
<td>Universal gas constant, dyn.cm/gr.K</td>
</tr>
<tr>
<td>$r$</td>
<td>Burning rate, gr/sec</td>
</tr>
<tr>
<td>$S$</td>
<td>Vector in Eq. 21</td>
</tr>
<tr>
<td>$T$</td>
<td>Temperature, K</td>
</tr>
<tr>
<td>$t$</td>
<td>Time, sec</td>
</tr>
<tr>
<td>$u$</td>
<td>Gas velocity, cm/sec</td>
</tr>
<tr>
<td>$x$</td>
<td>Axial distance, cm</td>
</tr>
<tr>
<td>$y$</td>
<td>Vertical coordinate, cm</td>
</tr>
</tbody>
</table>

Greece letters

<table>
<thead>
<tr>
<th>Symbol</th>
<th>Description</th>
</tr>
</thead>
<tbody>
<tr>
<td>$\alpha$</td>
<td>Thermal diffusivity, cm$^2$/sec</td>
</tr>
<tr>
<td>$\beta$</td>
<td>Constant of Eq. 29</td>
</tr>
<tr>
<td>$\gamma$</td>
<td>Ratio of heat capacities</td>
</tr>
<tr>
<td>$\epsilon$</td>
<td>Small number</td>
</tr>
<tr>
<td>$\epsilon_a$</td>
<td>Coefficient in artificial viscosity, Eq. 47</td>
</tr>
<tr>
<td>$\lambda_{pr}$</td>
<td>Conductivity of solid propellant, erg/cm.sec.K</td>
</tr>
<tr>
<td>$\pi$</td>
<td>Pi number</td>
</tr>
<tr>
<td>$\mu$</td>
<td>Viscosity, gr/cm.sec</td>
</tr>
</tbody>
</table>
υ  \hspace{1cm} \text{Kinematic viscosity, cm}^2/\text{sec}

τ \hspace{1cm} \text{Shear stress, dyn/cm}^2

Subscripts and superscripts
\begin{align*}
\text{e} & \quad \text{Nozzle exit} \\
\text{i} & \quad \text{Initial conditions} \\
\text{ig} & \quad \text{Heating gas} \\
\text{out} & \quad \text{Outlet} \\
\text{pr} & \quad \text{Solid propellant} \\
\text{n} & \quad \text{Value of quantities at time } n \\
\text{ps} & \quad \text{Propellant surface}
\end{align*}
References