

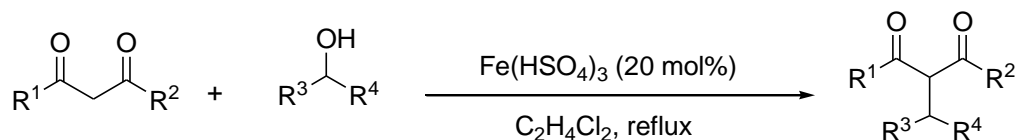
Fe(HSO₄)₃: An efficient, heterogeneous and reusable catalyst for alkylation of 1,3-dicarbonyl compounds

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The construction of carbon-carbon bond is a fundamental task in organic synthesis. Alkylation of active methylene compounds, such as β -diketones, β -diketoesters and malonic esters with alcohols is a useful transformation involving C-C bond formation.¹ However, because of the poor leaving ability of the hydroxyl group, excess amount of Brønsted or Lewis acids is required for this kind of transformation.² Various Lewis acids such as InCl₃,³ InBr₃,⁴ FeCl₃,⁵ M(OTf)₃,⁶ Lewis acid ruthenium complex,⁷ as well as montmorillonite,⁸ *p*-toluenesulfonic acid⁹ have been demonstrated to facilitate the dehydrative substitution of allylic and benzylic alcohols with 1,3-dicarbonyl compounds. Unfortunately many of these methods for the alkylation reaction has major or minor disadvantage such as long reaction times, low yields of the products, tedious work up procedures and use of the toxic solvents. Therefore, it is necessary to develop an efficient and convenient method for alkylation of 1,3-dicarbonyl compounds. Recently, ferric hydrogen sulfate Fe(HSO₄)₃ has emerged as a promising solid acid catalyst for acid catalyzed reactions. In this study we wish to report an efficient and convenient method for alkylation of a wide variety of 1,3-dicarbonyl compounds in the presence of Fe(HSO₄)₃ as catalyst.



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