

Spectroscopic and Structural Studies of Some New C(O)NHP(O) Containing Phosphorictriamides

A. Saneei *, M. Pourayoubi

Department of Chemistry, Ferdowsi University of Mashhad, Mashhad, 91779 Iran
(e-mail: anahid.saneei@yahoo.com)

New phosphoramidate compounds with formula XC(O)NHP(O)Y ($\text{Y} = \text{NHCH}_2\text{C}(\text{CH}_3)_2\text{CH}_2\text{NH}$ and $\text{X} = \text{CClF}_2$ (1) & CF_3 (2)) and X(O)NHP(O)Y_2 (for $\text{X} = \text{CClF}_2$, $\text{Y} = \text{NC}_4\text{H}_8$ (3), $\text{N}(\text{CH}_3)(\text{CH}_2\text{C}_6\text{H}_5)$ (4), $\text{NC}_4\text{H}_8\text{O}$ (5), $\text{NHCH}(\text{CH}_3)_2$ (6), $\text{NHC}_6\text{H}_4(4\text{-CH}_3)$ (7), $\text{NHC}_6\text{H}_{11}$ (8), NHC_6H_5 (9), $\text{NHCH}_2\text{C}_6\text{H}_5$ (10)) have been synthesized and characterized by IR, ^{19}F NMR, $^{31}\text{P}\{^1\text{H}\}$ NMR, ^{31}P NMR, ^1H NMR, ^{13}C NMR. Moreover, the structures of compounds 1, 2 and 3 were determined by single crystal X-ray determination. The $\text{P}=\text{O}$ and $\text{P}-\text{N}$ bond distances for these compounds are within the values characteristic of analogous phosphorus compound. Therefore, the bond angles values show that the P atom has a distorted tetrahedral configuration. The six-membered ring of the compounds 1 and 2 adopt a chair conformation and the $\text{P}=\text{O}$ unit exist in equatorial position.