

Synthesis, Spectroscopic Study and Crystal Structure of a New Amidophosphonate, (C₆H₅O)₂P(O)(NHCH(CH₃)(C₂H₅))

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The crystal structure of diphenyl (2-butylamido) phosphonate has been determined. This crystal belongs to the space group *P2₁/c*, and the asymmetric unit of the structure is composed of one complete molecule. The P atom has a distorted tetrahedral configuration with the O-P-O angle as the minimum bond angle at the P atom (97.74(18)°) and one of the O=P-O angles as the maximum angle (115.2(2)°). The oxygen atom of the P-O-C₆H₅ moiety may be ascribed with the *sp*² character, reflected in the P-O-C angles (120.8(3)° and 125.4(3)°). In the crystal structure, the molecules are aggregated through the N-H...O=P hydrogen bond (N1...O1 = 2.971(5) Å) in a one-dimensional chain along the *b* axis.

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Phosphorus-containing compounds play a vital role in different domains of chemistry,¹ because of their applications in medicine and pharmacology,^{2,3} coordination chemistry^{4,5} and biochemistry.⁶ Recently, crystal structures of some phosphoramides have been reported.^{7,8}

Here, we report on the synthesis and single-crystal X-ray determination of diphenyl (2-butylamido) phosphonate (I), with the chemical structure as shown in Fig. 1. For the synthesis of (I), a solution of 2-butylamine (0.37 g, 5 mmol) in dry acetonitrile (10 ml) was added to a solution of diphenyl phosphoryl chloride (0.67 g, 2.5 mmol) in the same solvent (20 ml) at 273 K. After stirring for 3 h, the solvent was removed *in vacuo* and the solid obtained was washed with distilled water. Single crystals of (I) were obtained from a solution of the product in methanol-heptane (4:1 v/v) after slow evaporation at room temperature. IR (KBr, cm⁻¹): 3242, 2972, 2928, 1742, 1592, 1490, 1375, 1250, 1203, 1151, 1071, 1025, 930, 767, 688. MS (70 eV): 305 (12) [M]⁺, 304 (35) [M-1]⁺, 289 (32) [M-CH₄]⁺, 275 (100) [M-C₂H₆]⁺, 182 (15) [M-C₂H₆-C₆H₅O]⁺, 95 (44) [C₆H₇O]⁺, 30 (90) [C₂H₆]⁺. ³¹P{¹H}-NMR (DMSO-*d*₆, 121.78 MHz, δ_{ppm}): 0.05 (s). ¹H-NMR: (DMSO-*d*₆, 300.85 MHz, δ_{ppm}): 0.76 (t, ³J_{HH} = 7.2 Hz, 3H, CH₃), 1.01 (d, ³J_{HH} = 6.6 Hz, 3H, CH₃), 1.34 (m, 2H, CH₂), 3.14 (m, 1H, CH), 5.70 (m, 1H, NH), 7.22 (m, 6H, Ar-H), 7.40 (m, 4H, Ar-H). ¹³C-NMR (DMSO-*d*₆, 75.66 MHz, δ_{ppm}): 10.77 (s), 22.73 (d, ³J_{PC} = 4.5 Hz), 31.24 (d, ³J_{PC} = 6.8 Hz), 49.68 (s), 120.56

(d, ³J_{PC} = 5.3 Hz), 120.58 (d, ³J_{PC} = 4.5 Hz), 125.05 (s), 130.15 (s), 151.32 (d, ²J_{PC} = 6.1 Hz), 151.34 (d, ²J_{PC} = 6.8 Hz). The ¹³C NMR spectrum is discussed in Supporting Information.

The crystallographic data and details of the X-ray analysis are presented in Table 1. Selected bond lengths and angles are given in Table 2. The asymmetric unit of structure (I) contains one complete molecule (Fig. 2). The P=O bond length (1.465(3) Å) is slightly longer than the P=O double bond length (1.45 Å)¹ and the P-N bond length (1.595(4) Å) is shorter than the standard P-N single bond length (1.77 Å).¹ The phosphorus

Table 1 Crystal and experimental data

Empirical formula: C ₁₆ H ₂₀ NO ₃ P	
Formula weight = 305.30	
Temperature = 100(2)K	
Crystal system: Monoclinic	Space group: <i>P2₁/c</i>
<i>a</i> = 13.277(2) Å	α = 90°
<i>b</i> = 5.2887(7) Å	β = 99.542(6)°
<i>c</i> = 21.962(5) Å	γ = 90°
<i>V</i> = 1520.8(5) Å ³	<i>Z</i> = 4
<i>D_x</i> = 1.333 g/cm ³	
Radiation: Mo <i>K</i> _α (λ = 0.71073 Å)	
μ(Mo <i>K</i> _α) = 0.190 mm ⁻¹	<i>F</i> (0 0 0) = 648
Crystal size = 0.1 × 0.08 × 0.08 mm ³	
No. of reflections collected = 2661	
No. of independent reflections = 2661	
θ range for data collection: 2.233 to 25.045°	
Data/restraints/parameters = 2661/15/193	
Goodness-of-fit on <i>F</i> ² = 1.074	
<i>R</i> indices <i>I</i> > 2σ(<i>I</i>): <i>R</i> ₁ = 0.0725, <i>wR</i> ₂ = 0.1485	
<i>R</i> indices (all data): <i>R</i> ₁ = 0.1093, <i>wR</i> ₂ = 0.1651	
(Δ/ <i>σ</i>) _{max} < 0.001	
(Δρ) _{max} = 1.240 eÅ ⁻³	(Δρ) _{min} = -0.577 eÅ ⁻³
Measurement: Bruker D8 Venture	
Program system: SHELXTL	
Structure determination: SHELXS ⁹	
CCDC deposition number: 1485675	

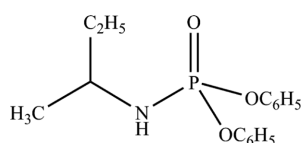


Fig. 1 Chemical structure of the title amidophosphonate.

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Table 2 Selected bond lengths [Å] and angles [°]

P1-O2	1.596(3)	O2-C5	1.397(6)
P1-O1	1.465(3)	N1-C1	1.484(6)
P1-N1	1.595(4)	O3-C11	1.406(5)
P1-O3	1.582(4)	C1-C2	1.651(8)
O1-P1-O2	115.2(2)	C1-N1-P1	127.8(3)
O1-P1-N1	115.0(2)	C11-O3-P1	125.4(3)
O1-P1-O3	114.6(2)	C10-C5-O2	119.1(4)
N1-P1-O2	106.7(2)	C6-C5-O2	119.1(4)
O3-P1-O2	97.74(18)	C16-C11-O3	122.6(4)
O3-P1-N1	105.8(2)	N1-C1-C2	107.4(5)
C5-O2-P1	120.8(3)	C3-C1-N1	112.4(5)

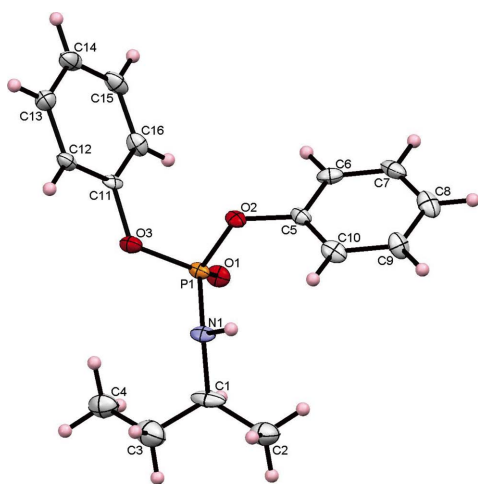


Fig. 2 Displacement ellipsoid plot (50% probability level) and the atom numbering scheme. H atoms are drawn as spheres of arbitrary radii.

atom has a distorted tetrahedral configuration. The bond angles around the phosphorus atom are in the range of $97.74(18)^\circ$ ($\angle O3-P1-O2$) to $115.2(2)^\circ$ ($\angle O1-P1-O2$). The oxygen atom of the P-O-C₆H₅ moiety may be ascribed with the *sp*² character, which is reflected to the C-O-P bond angles close to the *sp*² value of 120° ($\angle C5-O2-P1$: $120.8(3)^\circ$ and $\angle C11-O3-P1$: $125.4(3)^\circ$). The P-O bond lengths of the C-O-P fragments ($1.582(4)\text{Å}$ and $1.596(3)\text{Å}$) are shorter than the standard value well-known for the P-O single bond (1.64Å).¹ The dihedral angle between two phenyl rings is 54.20° .

The C1-N1-P1 angle of $127.8(3)^\circ$ is similar to the values reported for analogous structures with the (O)₂P(O)(NHC) skeleton.⁸ The NH group adopts a *gauche* orientation relative to the phosphoryl group (dihedral angle between H1N1P1 plane and O1P1N1 plane is 63.80°). The molecules are aggregated through the N1-H1...O1-P1 hydrogen bond (with $d(N1\cdots O1) = 2.971(5)\text{Å}$) in a one-dimensional chain along the *b* axis. The unit-cell packing is shown in Fig. 3 and hydrogen bonding data of the structure are presented in Table 3.

Acknowledgements

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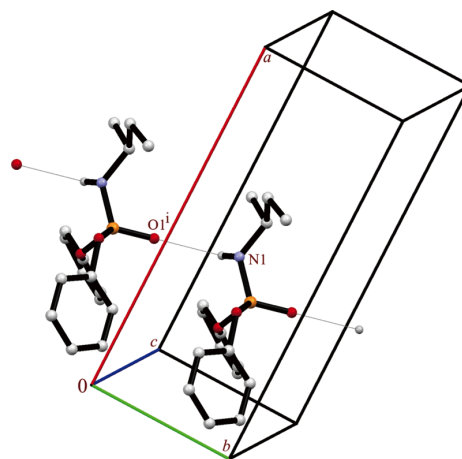
Fig. 3 Partial view of the crystal packing of (I), showing the linear arrangement built from the N-H...O=P hydrogen bonds (symmetry operation (i) $x, y-1, z$). The hydrogen bonds are shown as dotted lines. Only the H atoms involved in hydrogen bonding are shown.

Table 3 Hydrogen-bonding geometry (e.s.d. is given in parentheses)

D-H...A	D-H (Å)	H...A (Å)	D...A (Å)	$\angle D-H\cdots A$ (°)
N1-H1...O1 ⁱ	0.88	2.37	2.971(5)	126.2

Symmetry operation (i) $x, y-1, z$.

Supporting Information

This material is available free of charge on the Web at <http://www.jsac.or.jp/xraystruct/>.

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