

Abstract Preview of 'Diimine backbone' (C9W883)

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Preferred Symposium Reaction Engineering and Catalysis
Preferred Session Catalytic Reaction Engineering
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Invited Speaker -
Learning Outcomes Steric and electronic effects of backbone structure on catalyst behavior in
1- hexene polymerization

Preview of your abstract

Polymerization of 1-hexene using binuclear LTM catalysts **M. Khoshsefat** <Mostafa.khoshsefat@yahoo.com>, **A. Dechal** <A.Dechal@ippi.ac.ir>, **S. Ahmadjo** <S.Ahmadjo@ippi.ac.ir> and **M.M. Mortazavi** <M.Mortazavi@ippi.ac.ir>, Department of Catalyst, Iran Polymer and Petrochemical Institute (IPPI), P.O. Box 14965/115, Tehran,Iran; **G.H. Zohuri** <Zohuri@um.ac.ir>, Department of Chemistry, Ferdowsi University of Mashhad, P.O. Box: 91775,Mashhad, Iran

Two binuclear late transition metal (LTM) catalysts (A=2,4,6-trimethyl-C₆H₂-N=C-(C₁₀H₆) -C=N-C₆(CH₃)₄-N=C-(C₁₀H₆)-C=N-2,4,6-trimethyl-C₆H₂)(NiBr₂)₂ and B=2,4,6-trimethyl-C₆H₂-N=C-(C₂H₆)-C=N-C₆(CH₃)₄-N=C-(C₂H₆)-C=N-2,4,6-trimethyl-C₆H₂)(NiBr₂)₂) were used in 1-hexene homopolymerization. The experiments carried out using A/EASC and B/EASC catalytic systems at the optimum conditions of each catalyst. The productivity of each catalyst, viscosity average molecular weight (M_v) and polydispersity index (PDI) of obtained poly(1-hexene) were compared. The activity of A was higher than B due to presence of acenaphthene group on C-C bond through the increasing of steric hindrance, decreasing of N-Ni-N bond angle leading to block axial sites by methyl groups and high cooperative effect of metal centers in the A. Moreover, the M_v of resulted poly(1-hexene) in presence of catalyst (A) was 2.32×10⁵ g. mol⁻¹ and PDI=2.11 which were higher and broader than B catalyst (M_v=1.56×10⁵ g.mol⁻¹, PDI=1.94). These results can be attributed to steric and electronic effects of backbone which can control the rate of propagation to chain transfer reactions.

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